# **Permeability of a Cell Junction During Intracellular Injection of Divalent Cations**

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*Summary.* Divalent cations are microinjected into *Chironomus* salivary gland cells while the cell-to-cell passage of fluorescein (330 dalton) and electrical coupling are monitored. Injections of Ca and Mg that substantially depolarize the cells produce block or marked slowing fluorescein passage, accompanied by electrical uncoupling. Injections of Ca, Mg or Sr that cause little depolarization, and presumably smaller elevation of divalent cation concentration in the cytoplasm, produce block or marked slowing of fluorescein passage with little or no detectable electrical uncoupling. This partial uncoupling may reflect total closure of a fraction of the channels in junctional membrane or partial closure of all channels.

In a wide variety of tissues, the intracellular space is made continuous by cell membrane junctions that provide a path for direct cell-to-cell diffusion of molecules of a certain size range (Loewenstein, 1966, 1975 $c$ ; Furshpan & Potter, 1968). The permeability of these junctions is markedly reduced when  $Ca^{2+}$  flux into the cells is excessively high, as in the case of damage to (nonjunctional) cell membrane in Ca-medium (D616ze, 1964, 1970, 1975; Loewenstein, 1966; Loewenstein, Nakas & Socolar, 1967; Oliveira-Castro & Loewenstein, 1971), or when the cytoplasmic  $Ca^{2+}$ concentration in the intact cells is elevated in several experimental conditions (Politoff, Socolar & Loewenstein, 1969; Rose & Loewenstein, 1971). This has led to the hypothesis that the junctional membrane permeability is determined by the cytoplasmic  $Ca^{2+}$  concentration (Loewenstein, 1967). In the present work we have attempted to test this hypothesis by injecting  $Ca^{2+}$  and other divalent cations into the cell, while monitoring cell-to-cell passage of fluorescein and electrical coupling.

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Ftuorescein passage, indeed, was found to be depressed; but electrical coupling was depressed only in certain of the  $Ca^{2+}$  injections, namely in those producing depolarization. This raised the question of whether the injected  $Ca^{2+}$  had spread far enough inside the still coupled cells to reach a critical concentration at the junctions. To resolve this crucial point, we postponed publication of this work (performed in 1971/1972) until techniques for monitoring the intracellular free Ca concentration could be developed. This has now been done and the companion paper describes the complementary results. These, we believe, provide a full explanation of the present findings and show that junctional permeability indeed depends on the cytoplasmic free  $Ca^{2+}$  concentration in the junctional locale (Rose & Loewenstein, 1976).

#### **Materials and Methods**

Salivary glands of *Chironomus thummi* were isolated in physiological medium (Politoff *et al.,* 1969). Electrical coupling was measured by pulsing inward current  $(0.4 \times 10^{-7}$  or  $1 \times 10^{-7}$  A; 0.2 sec duration;  $\sim 12$ /min) into a cell (I) and measuring the resulting steadystate displacements of membrane voltage  $(V)$  in this cell and a contiguous one *(II) (Fig. 1, inset*). The ratio  $V_{II}/V_I$  (*coupling ratio*) provided a convenient index of the coupling (Loewenstein & Kanno, 1964).

Divalent cations (Ca,  $8-30$  mm; Mg,  $15-30$  mm; or Sr,  $3-9$  mm) were injected into cell *II* as Cl-salt solutions, made isotonic (260 mOsm) by addition of KC1 or of a mixture of KC1  $(2/3)$  of the required osmolar concentration) and sucrose  $(1/3)$ . The divalent cation solutions also contained Na fluorescein  $(2-6 \text{ mM})$  to make visible the outflow from the injection micropipette and to probe junctional permeability (Loewenstein  $&$  Kanno, 1964). Pressure pulses, variable in strength, duration and frequency, were applied to the injection micropipettes while electrical coupling and membrane potential were measured. To avoid cell damage by excessive pressure, the injections were begun with pulses, 10/min, of increasing pressure and duration up to injection threshold as evidenced by the sudden appearance of a bright intracellular fluorescence. The pressure pulses were then routinely continued for 20-60 sec in an effort to inject a sufficient volume; but the injections seemed generally effectively limited to the first few pulses, as the divalent cation-containing pipettes tended to clog. The fluorescent pipette meniscus was photographed immediately before and after injection (e.g.  $M_1$  and  $M_2$ , Fig. 1) and the injected volume was calculated as that of the corresponding truncated cone or cylinder in the pipette. The height of the cone or cylinder was determined from the distance between the meniscus projections on the horizontal plane as measured on the photographs and from the angle of the micropipette with this plane. It is difficult to assess an error on the volume measurements, but there is no evident argument for a substantial systematic overestimate: evaporation was negligible. However, some underestimate due to a rise in fluid level by capillarity after injection was possible, although controls with pipettes inside the cells showed httle, if any, sign of such a rise.

In the series of control injections in which K substituted for the divalent cations, the injected solutions contained besides Na-fluorescein  $(2-6 \text{ mm})$ , either 130 mm KCl  $(4 \text{ inject})$ tions), 45 mm KCl (1 injection), or a mixture of 45 mm KCl and 170 mm sucrose (3 injections).

Injected cation	$n^a$	Amount injected <sup>b</sup> pmole	Coupling ratio $V_{II}/V_I$ <sup>c</sup>		Cell-to-cell flux of	$\Delta$ Membrane potential <sup>d</sup> $\frac{6}{6}$	
			before injection	1 min after injection	fluorescein	Cell I	Cell II
Group A							
K	8	$32 - 133$	$0.92 + 0.02$	$0.90 + 0.03$	normal	$-13.9 + 2.7$	$-15.9+3.9$
Ca	10	$2 - 63$	$0.90 + 0.02$	$0.86 + 0.02$	blocked <sup>e</sup> or slowed	$-7.7 + 3$	$-12.5+3.4$
Mg	6	$5 - 18$	$0.92 + 0.02$	$0.88 + 0.04$	slowed		$-11.9+6$ $-8.3+6.1$
Sr	$\overline{4}$	$1.5 - 12$	$0.95 + 0.03$	$0.91 \pm 0.02$	slowed		$-10.3 \pm 5.9$ $-11.9 \pm 6.2$
$Group\ B$							
K	$\mathbf{1}$	66	0.95	0.44	blocked	$-31.3$	$-59$
Ca	$\leq$	$6 - 132$	$0.89 + 0.02$	$0.26 + 0.11$	blocked <sup>f</sup> or slowed	$-42.4 + 8.9$	$-76.5 \pm 10$
Mg	1	33	0.84	0.31	blocked	$\theta$	$-64.3$

Table 1. Effects of cation injection on coupling and membrane potential

<sup>a</sup> Number of cases.

<sup>b</sup> Ranges: lowest and highest value in each group.

~ Mean with SE.

 $d$  The change in the membrane potential (at zero current) 1 min after start of injection; mean membrane potentials and se in 32 experiments were  $43 \text{ mV} \pm 19 \text{ } (E_t)$  and  $41.9 \pm 19 \text{ } (E_{tt})$ . e In 5 cases.

f In 4 cases.

#### **Results**

## *Electrical Coupling*

The effects of the divalent cation injections are summarized in Table 1. The results on electrical coupling fell into two groups, depending on the degree of depolarization associated with the injection. In one group  $(A)$ , in which the average depolarization did not exceed  $12\%$  of the initial membrane potential, no significant decrease in coupling ratio *(uncoupling)*  was detectable over  $3-12$  min of continuous coupling measurement (Figs. 1-3). In the other group  $(B)$ , in which the depolarization was substantial (in all cases  $>30\frac{\%}{6}$ ; in most  $\geq 45\frac{\%}{6}$ ), the cells uncoupled. The coupling ratios decreased to a fraction of their initial values (in the Ca injections the ratios decreased on the average by a factor of 3.4) or to below detectable level. In a typical example of this group, shown in Fig. 4A, the membrane potential  $(E_{II})$  of the injected cell and the transfer voltage  $V_{II}$  fell abruptly at the onset of the Ca injection. Subsequently  $V_I$ rose progressively while  $V_{II}$  stayed near zero, showing conclusively that



Fig. 1. Partial uncoupling by Ca (block of fluorescein passage). Membrane potential, electrical coupling and cell-to-cell flow of fluorescein during intracellular injection of a solution containing Ca and fluorescein as a tracer. Upper right inset: diagram of cell chain with current-passing electrode (i), recording electrodes  $(V_I, V_{II})$  and injection micropipette in cells I and II. (A) values of the membrane potentials  $(E_I, E_{II})$  showing also the steady-state displacements  $(V_I, V_{II})$  during test current pulses  $(i=0.4 \times 10^{-7} \text{ A})$  and the coupling ratios  $(V_{II}/V_I)$ , sampled from a continuous record. Bar starting at time 0 marks the injection of 4 pmole Ca in cell II. The transient depolarization near  $-1$  min was caused by insertion of the injection micropipette (white arrow); typically the membrane "seals" in the following min. (B) Cell-to-cell flow of fluorescein. Darkfield micrographs of the fluorescence of the living cells after injection in cell II. At bottom, the times after injection the photographs were taken. Although electrical coupling was maintained (group  $A$  in Table 1), there was no detectable fluorescein diffusion beyond the boundaries of cell II. Superposed at tops are photographs of the fluorescent solution in the micropipette showing the meniscus  $M_1$ ,  $M_2$ before and after the injection. To the right is a tracing of the outlines of cells I and II and one neighbor. Calibration in this and subsequent figures,  $100 \mu$ 



Fig. 2. Partial uncoupling by Ca (slowing of fluorescein passage). Set-up and notation in this and subsequent figures as in Fig. 1. Test current pulses  $0.4 \times 10^{-7}$  A. Bar starting at time 0 marks the injection of an unknown amount of Ca.  $V_{II}/V_I$  fell during the injection, but is close to control values between 1 and 4 min (group-A experiment). 1 min after the injection, fluorescein is still restricted to the injected cell  $(II)$ . At 5 min a distinct fluorescence appears in the two adjacent cells  $(I \text{ and } III)$ , but the discontinuity in brightness at the boundaries of cell II indicates a constraint on fluorescein diffusion



Fig. 3. Partial uncoupling by Mg (slowing of fluorescein passage). Test current pulses  $1 \times 10^{-7}$  A. Bar starting at time 0 marks the injection of 9 pmole Mg in cell II. Cells remain electrically coupled (group-A experiment). Slight passage of fluorescein to adjacent cells  $(I, III)$  is detectable at 4 min

the junctional membrane conductance had markedly fallen (See Loewenstein et al., 1967).

## Fluorescein Passage

Cell-to-cell passage of fluorescein was either reduced below detectable magnitude (Figs. 1 and 4) or markedly slowed (Figs. 2 and 3) in all cases of divalent cation injection, including those of group  $A$  in which the cells



Fig. 4. General uncoupling by Ca. Test current pulse  $0.4 \times 10^{-7}$  A. Bar at time 0 signals. the sudden injection of an unknown amount of Ca in cell  $II$ . The injection is followed by a substantial depolarization of cell  $II$  and by electrical uncoupling as shown by the decrease of  $V_{II}$  and the rise of  $V_I$ , and by the divergence of  $E_I$  and  $E_{II}$  (group-B experiment). Fluorescein is confined to the injected cell



Fig. 5. Control injections of K. White arrow at start of plot indicates insertion of the injection micropipette in cell II. Bar marks the injection period of a solution containing 130 mm KCl and 2.5 mm fluorescein. (A) values of E, V and  $V_{II}/V_I$  ( $i = 1 \times 10^{-7}$  A). (B) darkfield micrograph from the same experiment showing the cell-to-cell passage of fluorescein. (C) darkfield micrograph from a similar experiment showing uniform distribution of the injected fluorescein 1 min after injection

stayed well-coupled electrically. However, the intracellular diffusion of fluorescein appeared unchanged; the spatial extent and the rate of fluorescein diffusion within the injected cell was not noticeably altered. In groups  $A$  and  $B$ , fluorescein diffusion appeared always to be restricted to the injected cell for at least the first 2 min after injection (with Ca

injections, for at least the first 4 min). In the next 5 min, a rather weak fluorescence appeared in 1 to 4 adjacent cells in 50% of the experiments of group A and 15% of group B, but there was then at all times an obvious discontinuity in brightness at the junctional cell boundaries (Figs. 2 and 3). This discontinuity and the slowness of the fluorescence spread across the junction contrasted sharply with the pattern of cell-to-cell spread of fluorescein when divalent cations were absent from the injected solution. Then the fluorescein was seen to cross the junctional boundary within a few seconds of the injection, and the fluorescence intensity appeared continuous across that boundary, falling off gradually with distance from the injection site (Loewenstein & Kanno, 1964; Oliveira-Castro  $&$  Loewenstein, 1971). Thus, in the presence of divalent cations, junctional diffusion of the fluorescein molecule seems to be severely constrained.

## *K Cl Injection*

Electrical coupling was generally normal and fluorescein passage unconstrained when K substituted for the divalent cations in the injected solution (except for a small amount of Na, K was then the only injected cation; *see* Materials and Methods). In 8 experiments in which little depolarization occurred during the injection, the electrical coupling remained unchanged (Table 1) and fluorescein spread visibly to the adjacent cells within a few seconds, distributing itself rather continuously across the cell boundaries (Fig. 5).

In the one experiment exhibiting a marked depolarization, the coupling ratio fell from 0.95 to 0.44 and fluorescein passage was blocked. This effect may be attributed to  $Ca^{2+}$  released from intracellular stores, because it was associated with substantial depolarization which is generally not produced by K injection. As will be shown in the following paper, injection of KCl in sufficient amount can cause the release of  $Ca^{2+}$  from intracellular stores, and this release is invariably accompanied by depolarization because an elevation of cytoplasmic  $\lceil Ca^{2+} \rceil$  produces an increase in nonjunctional membrane conductance (Rose  $\&$  Loewenstein, 1976).

#### **Discussion**

## *General Uncoupling*

*Uncoupling for small inorganic ions.* The discrepant results of the divalent cation injections on electrical coupling in the experiment groups  $A$  and  $B$  were at first puzzling: electrical uncoupling ensued

only when the injection produced a substantial depolarization. Taken alone, this result seemed to suggest that the divalent cations had no unique action on junctional membrane permeability and that the depolarization might be a primary cause of the uncoupling, particularly since depolarization by outward current can induce uncoupling (Socolar & Politoff, 1971) and repolarization by inward current can reverse it under certain conditions (Rose, 1970). Yet the results of other experiments in which various cations were made to enter the cell from an infinite external reservoir, swamping the cell interior, indicated a rather specific action of divalent cations on junctional permeability (Oliveira-Castro & Loewenstein, 1971). Moreover, experiments in which the cells were depolarized by exposure to extracellular medium of high K concentration showed that depolarization does not suffice for uncoupling (Rose & Loewenstein, 1971).

All these results could, however, be explained if we assume (i) that the divalent cations in the experiments of group A somehow did not reach the junctions and *(ii)* that depolarization, at least under some conditions, leads to substantial elevation of cytoplasmic  $Ca^{2+}$  concentration.

Notion i may at first glance seem far-fetched because the injected Ca and Mg quantities were high (Table 1). The intracellular concentrations at the junctions should indeed have been sufficient for uncoupling were the cells passive receptacles, and the divalent cations, like other ions, free to diffuse in the cytoplasm<sup>1</sup>. However, in respect to  $Ca^{2+}$ ,  $Mg^{2+}$ , and  $Sr^{2+}$ , the cells are all but passive. These ions  $-$  and  $Ca^{2+}$  particularly fast- are sequestered by mitochondria acting like high-capacity sinks *(cf* Lehninger, Carafoli & Rossi, 1967; Pressman, 1970). Moreover, these ions are bound by cytoplasmic protein (Baker & Schaepfler, 1976). Therefore, in spreading through the cytoplasm, the concentrations of the injected divalent cations fall off more steeply with distance than in simple diffusion (Harris, 1957; Hodgkin & Keynes, 1957; Blaustein & Hodgkin, 1969; Baker & Crawford, 1972). In fact, it has just been shown (Rose & Loewenstein, 1975a, b) that the Ca<sup>2+</sup> diffusion restriction in *Chironomus* salivary gland cells is so severe that an increase in the cytoplasmic Ca<sup>2+</sup> concentration ([Ca<sup>2+</sup>]<sub>i</sub>), produced by Ca injection similar to those we have used here, is restricted to the immediate vicinity of the

<sup>1</sup> With a diffusion constant of, for instance,  $1.2 \times 10^{-6}$  cm<sup>2</sup>s<sup>-1</sup> (Hodgkin & Keynes, 1957) and a rate constant for nonjunctional Ca extrusion not in excess of  $2 \times 10^{-3}$  min<sup>-1</sup> (Baker, 1972), it can be shown with the aid of standard equations for random-walk diffusion that the Ca concentration at the junction would be of the order of  $10^{-3} - 10^{-2}$  M in the various trials, 1 min after injection, well above the uncoupling threshold.

injection micropipette (even when the Ca in the injection is buffered). When the injection was made into the cell center, raising the  $\lceil Ca^{2+} \rceil$ . within a radius of 5  $\mu$  above 10<sup>-4</sup> M, the [Ca<sup>2+</sup>]<sub>i</sub> at the cell junction - on the average 50  $\mu$  away from the center – stayed below 10<sup>-5</sup> M, the threshold concentration for electrical uncoupling. This was so unless  $\lceil Ca^{2+} \rceil$ , at the cell center was seen to rise particularly rapidly during injection.

Thus, it seemed indeed possible that the injected divalent cations in the experiments of group  $A$  did not reach the cell junction in concentrations sufficient for uncoupling. The factor determining the uncoupling in group  $\hat{B}$  may simply have been the degree of eccentricity of the injection site, the sequestering rate, or the degree of saturation of the sequestering capacity. To resolve this crucial point, one needs to know the divalent cation concentration at the junction during the injection. The means for obtaining this information were developed over the past three years, and the experiments are described in the companion paper (Rose  $\&$ Loewenstein, 1976). The results show that uncoupling occurs when the  $[Ca^{2+}]$  in the junctional domain rises above about  $5 \times 10^{-5}$  M and only then. They reveal furthermore that the depolarization is a byproduct of a sufficient  $\lceil Ca^{2+} \rceil$ , elevation at nonjunctional membrane.

*Uncoupling for fluorescein.* Cell-to-cell passage of fluorescein was hindered in all cases of electrical uncoupling (group B, Table 1), as is to be expected if the small inorganic ions, the carriers of the electrical current, and the larger fluorescein anion (300 dalton) pass through the same junctional membrane channels *(see* Loewenstein, 1966 and 1975a, for a definition of the junctional channel unit). That in some Ca injections fluorescein passage was slowed, whereas in others it was sensibly blocked, may be a reflection of the number of junctional channels closed by  $Ca^{2+}$ action. It may also reflect differences between the rates with which fluorescein and Ca reach sufficient concentration at the junction; that is, slowing, rather than blockage, ensues when fluorescein gets to the junction before  $Ca^{2+}$  attains a concentration there sufficient for channel closure. This is a distinct possibility, because the slowing in the group B experiments occurred in the case with the largest injection volume (132 pmoles) and because fluorescein, unlike  $Ca^{2+}$ , is not avidly sequestered in the cytoplasm (see Fig. 2 of Rose & Loewenstein, 1975b).

## *Selective Uncoupling?*

Of great interest is the finding that fluorescein celI-to-cell passage is hindered by divalent cation injection even when the cells stay coupled electrically (group A, Table 1). This may mean that the patency or the

charge of the junctional membrane channels changes gradually with divalent cation concentrations; that is, the effective channel bore is reduced even at the low cytoplasmic divalent cation concentrations in the junctional locale in the group  $A$  experiments. A sufficient reduction in the actual patency of the channel would be more limiting for the 300-dalton fluorescein than for the much smaller inorganic ions; or a shift toward negativity of channel fixed charges (resulting from a conformational change) could discriminate against the fluorescein anion. An alternative interpretation is that the effect of the divalent cations is an all-or-none channel closure, the proportion of the channels closing being a direct function of divalent cation concentration. The first interpretation, that of a truly selective uncoupling according to molecular size or charge, has the more interesting physiological implications, but the second is not excluded because the sensitivity of the fluorescein method for detecting closure of a fraction of the channels may be greater than that of the electrical method. The (electrical) coupling ratio is relatively insensitive to the proportion of channels open, except in the low range. For instance, if we consider the simple case of a coupled symmetrical cell pair, where the coupling ratio is given by  $V_2/V_1 = r_o/(r_o + r_c)$ , and take the nonjunctional membrane resistance  $(r<sub>o</sub>)$  to be 10 M $\Omega$  and the junctional membrane resistance  $(r_c)$  to be 50 k $\Omega$  (Loewenstein *et al., 1967*; Rose, 1971), then closure of 95% of the junctional channels would give only a 10% reduction in coupling ratio.

In view of the uncertainty about which of these two explanations applies here, we will refer to the phenomenon as *partial* uncoupling.

A further possibility, that the fluorescein and inorganic ions take altogether different routes seems unlikely in view of several other findings: (i) A broad variety of colorant and fluorescent tracer substances which do not permeate nonjunctional membrane pass, like fluorescein, rapidly from cell to cell (Kanno & Loewenstein, 1966; cf. Loewenstein, 1975 c; I. Simpson, B. Rose, H. Wiegandt & W. R. Loewenstein, *unpublished; see also* Payton, Bennett & Pappas, 1969; Johnson & Sheridan, 1971; Imanaga, 1974). *(ii)* Cell-to-cell passage of fluorescein is blocked whenever the junctional passage of small inorganic ions is blocked by experimental elevation of  $\lceil Ca^{2+} \rceil$ , (Loewenstein, 1967; Oliveira-Castro & Loewenstein, 1971; Rose & Loewenstein, 1971). *(iii)* Certain cell strains which are genetically incapable of junctional passage of small inorganic ions are also incapable of fluorescein passage. When these strains are rendered capable of fluorescein passage by genetic manipulation, they become also capable of small-ion passage (Azarnia, Larsen & Loewenstein, 1974).

One further possibility, that of selective uncoupling of the giant cell junction will be discussed. The *giant* salivary gland cells, the cells which we inject into and record from, are coupled to each other directly via their common junctions and indirectly via their junctions with 2 or 3 *fiat*  cells spanning the lumen. Because the flat cells are very thin, it is difficult, except under favorable conditions, to demonstrate the presence of fluorescein inside them; yet they provide a path for close electrical coupling between the giant cells (Rose, 1971). Hence, if the giant cell junctions were blocked to the passage of both inorganic and fluorescein ions in the group-A experiments, and the electrical coupling were mediated by the flat cell junctions alone, fluorescein cell-to-cell passage might not have been seen in the flat cells. However, the possibility of a selective uncoupling of this kind seems unlikely because with the quantities (and rates) of the fluorescein injections in the group- $A$  experiments the fluorescein should have become visible in the adjacent giant cells had the flat cell junctions been open to the passage of this molecule. Although the fluorescein is generally not visible in the flat cells themselves, these cells provide normally a good path for fluorescein passage between giant cells; fluorescein transfer between giant cells via flat cell junctions has been demonstrated with fluorescein injections comparable to the present ones, even where the giant cells were farther apart (Rose, *personal communication).* Besides, there are no obvious reasons why the giant cell junction should behave differently from the flat cell junction in the conditions of the group-A experiments. Both kind of junctions are anatomically and functionally similar (Rose, 1971), and their topography is such that in any given divalent cation injection, regardless of its centricity, flat cell junctions are as close or closer to the injection site than one of the two giant cell junctions of the injected cell.

General uncoupling is shown in the following paper to require a [Ca<sup>2+</sup>], in the junctional domain  $\gtrsim 5 \times 10^{-5}$  M (Rose & Loewenstein, 1976). Partial uncoupling thus must take place at even lower  $[Ca^{2+}]_{i}$ , presumably in the range  $10^{-7}$  to  $5 \times 10^{-5}$  M. This points up a danger in the studies of junctional permeability which was hitherto not apparent. The  $\lceil Ca^{2+} \rceil$ , normally  $\lt 10^{-7}$  (Baker, 1972), can rise to this range as the result of experimental manipulations;  $Ca^{2+}$  leakage from the medium through an insufficiently sealed membrane region of microelectrode insertion, or depression of cell energy metabolism (on which the  $\lceil Ca^{2+} \rceil$ ultimately depends) by an inadequate external medium, are only two of several possible causes of artifactual uncoupling.

In this light, we shall discuss a recent report by Van Venrooij, Hax,

Schouten, Denier van der Gon and van der Vorst (1975) that the junctional permeability for fluorescein and other similar-sized fluorescent molecules is low in salivary gland cells of *Drosophila hydeii* (end third-instar stage). Their sample photographs of intracellular fluorescence distribution on which the permeability calculations are based show some fluorescence transfer, but with sharp discontinuities in intensity at the cell boundaries. With molecules of the size of fluorescein, this is typical of partial uncoupling in *Chironomus* salivary gland cells with maintained electrical coupling *(see* Results). Furthermore, in comparing the results of Van Venrooij *et al.,* with those obtained with salivary gland cells *of Drosophilafulvimacula flavorepleta* (early and mid third-instar stage) (Loewenstein & Kanno, 1964), which are quite similar in dimension, configuration and grouping to those of *D. hydeii,* one is struck by the difference in the extent of the fluorescein spread along the major gland axis. In the experiments on *D. hydeii,* the injected fluorescein did not spread detectably beyond two cell junctions, whereas in those on *D. fulvimacuIa* it spreads, even with short injections, through  $>$  20 junctions (in fact, often throughout nearly the entire gland). Thus, Van Venrooij *et al.* may have worked unwittingly on a partially uncoupled cell system. The uncoupling may have been an experimental artifact or an uncoupling occurring naturally during development between the mid and end third-instar stage.

Slack and Palmer (1969) made the interesting discovery that there is no detectable fluorescein transfer between electrically coupled cells of the early *Xenopus* embryo. This has been observed also in various other embryos (Bennett, Pappas & Spira, 1972; Tupper & Saunders, 1972; but *see also* Sheridan, 1971). This partial uncoupling may be an important physiological mechanism in embryonic development, namely in the establishment and fixation of cell differentiation (Loewenstein, 1968; 1975 b). Since it bears at least a superficial resemblance with the present partial uncoupling produced by  $Ca^{2+}$  injection, it may be rewarding to invest future effort in studying the  $\lceil Ca^{2+} \rceil$ , in such embryo cells.

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#### **References**

Azarnia, R., Larsen, W., Eoewenstein, W.R. 1974. The membrane junctions in communicating and non-communicating cells, their hybrids and segregants. Proc. Nat. Acad. Sci. *USA* 71:880

- Baker, R E 1972. Transport and metabolism of calcium ions in nerve. *Prog. Biophys. Biophys. Chem.* 24 : 177
- Baker, P.F., Crawford, A.C. 1972. Mobility and transport of magnesium in squid giant axons. *J. Physiol. (London)* 227:855
- Baker, E E, Schaepfler, 1976. Calcium uptake by axoplasm extruded from giant axons of *Loligo. J. Physiol. (London) (in press)*
- Bennett, M.V.L., Pappas, G.D., Spira, M.E. 1972. Properties of electrotonic junctions between embryonic cells of *Fundulus. Dev. Biol.* 29:419
- Blaustein, M. E, Hodgkin, A.L. 1969. The effect of cyanide on the efflux of calcium from squid axons. *J. Physiol. (London)* 200:497
- D616ze, J. 1964. Calcium ions and the healing-over of heart fibers. *In:* Electrophysiology of the Heart. B. Taccardi and C. Marchetti, editors, p. 147. Pergamon, Oxford
- D616ze, J. 1970. The recovery of resting potential and input resistance in sheep heart injured by knife or laser. *J. Physiol. (London)* 208: 547
- D616ze, J. 1975. The site of healing over after a local injury in the heart. *In:* Recent Advances in Studies on Cardiac Structure and Metabolism. A. Fleckenstein and N. S. Dhalla, editors. Vol. 5, p. 223. University Park Press, Baltimore
- Furshpan, E.J., Potter, D.D. 1968. Low resistance junctions between cells in embryos and tissue culture. *Curr. Top. Dev. Biol.* 3:95
- Harris, E.J. 1957. The output of <sup>45</sup>Ca from frog muscle. *Biochim. Biophys. Acta* 23:80
- Hodgkin, A.L., Keynes, R.D. 1957. Movements of labelled calcium in squid giant axons. *J. Physiol. (London)* 138:253
- Imanaga, I, 1974. Cell-to-cell diffusion of Procion Yellow in sheep and calf Purkinje fibers. *J. Membrane Biol.* 16:381
- Johnson, R., Sheridan, J.D. 1971. Junctions between cancer cells in culture: Ultrastructure and permeability. *Science* 174:717
- Kanno, Y., Loewenstein, W. R. 1966. Cell-to-cell passage of large molecules. *Nature (London)*  212: 629
- Lehninger A.L., Carafoli, E., Rossi, C.J. 1967. Energy-linked ion movements in mitochondrial systems. *Adv. EnzymoI.* 29-259
- Loewenstein, W.R. 1966. Permeability of membrane junctions. *Ann. N. Y Acad. Sci.* 137:441
- Loewenstein, W.R. 1967. Cell surface membranes in close contact. Role of calcium and magnesium ions. *J. Colloid Interface Sci.* 25:34
- Loewenstein, W.R. 1968. Communication through cell junctions. Implications in growth control and differentiation. *Dev. Biol.* 19 (Suppl. 2): 151
- Loewenstein, W.R. 1975a. Cellular communication by permeable junctions. *In:* Cell Membranes: Biochemistry, Cell Biology and Pathology. G. Weissman and R. Claiborne, editors, p. 105. HP Publishing Co., Inc., NewYork
- Loewenstein, W.R. 1975b. Direct cell-to-cell communication. *In:* From Theoretical Physics to Biology. M. Marois, editor, p. 101. North Holland Publishing Company, Amsterdam

Loewenstein, W.R. 1975 c. Permeable junctions. *Cold Spring Harbor Syrup. Quant. Biol.* 40:49

- Loewenstein, W. R., Kanno, Y. 1964. Studies on an epithelial (gland) cell junction. I. Modifications of surface membrane permeability. *J. Cell Biol.* 22:565
- Loewenstein, W.R., Nakas, M., Socolar, S.J. 1967. Junctional membrane uncoupling. Permeability transformations at a cell membrane junction. *J. Gen. Physiol.* 50:1867
- Oliveira-Castro, G.M., Loewenstein, W.R. 1971. Junctional membrane permeability: Effects of divalent cations. *J. Membrane Biol.* 5:51
- Payton, B.W., Bennet, M.V., Pappas, G.D. 1969. Permeability and structure of junctional membranes at an electrotonical synapse. *Science* 166:1641
- Politoff, A.L., Socolar, S.J., Loewenstein, W.R. 1969. Permeability of a cell membrane junction. Dependence on energy metabolism. *J. Gen. Physiol.* 53:498
- Pressman, B.C. 1970. Energy-linked transport in mitochondria. *In:* Membranes of Mitochondria and Chloroplasts. E. Racker, editor. Van Nostrand Reinhold Co., New York
- Rose, B. 1970. Junctional membrane permeability: Restoration by repolarizing current. *Science* 169:607
- Rose, B., Loewenstein, W.R. 1971. Junctional membrane permeability: Depression by substitution of Li for extracellular Na, and by long-term lack of Ca and Mg; restoration by cell repolarization. *J. Membrane Biol.* 5:20
- Rose, B., Loewenstein, W.R. 1975a. Permeability of cell junction depends on local cytoplasmic calcium activity. *Nature (London)* 254:250
- Rose, B., Loewenstein, W.R. 1975b. Calcium ion distribution visualized by aequorin: Diffusion in the cytosol is restricted due to energized sequestering. *Science.* 190:1204
- Rose, B., Loewenstein, W.R. 1976. Permeability of a cell junction and the local cytoplasmic free ionized calcium concentration: A study with aequorin. *J. Membrane Biol.* 28:87
- Sheridan, J. 1971. Dye movement and low resistance junctions between reaggregated embryonic cells. *Dev. Biol.* 26:627
- Skack, C., Palmer, J. R 1969. The permeability of intercellular junctions in the early embryo of *Xenopus laevis,* studied with fluorescent tracer. *Exp. Ceil Res.* 55:416
- Socolar, S.J., Politoff, A.L. 1971. Uncoupling cell junctions of a glandular epithelium by depolarizing current. *Science* 172:492
- Tupper, J., Saunders, J.W. 1972. Intercellular permeability in the early *Asterias* embryo. Dev. Biol. 27:546
- Van Venrooij, Hax, W.M.A., Schouten, V. J.A., Denier van der Gon, J.J., van der Vorst, H.A. 1975. Absence of cell communication for fluorescein and dansylated aminoacids in an electrotonic coupled cell system. *Biochim. Biophys. Acta* 394:620